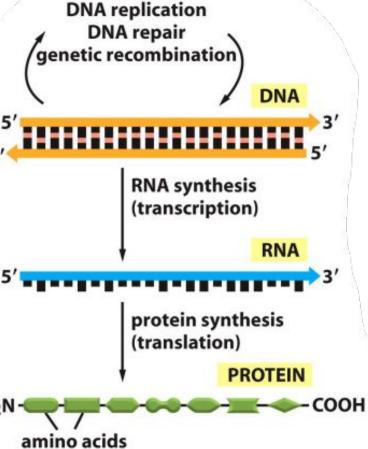
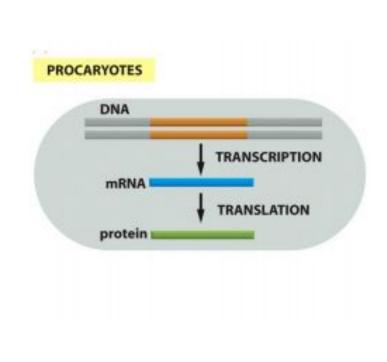
# **Bi 8 Midterm Review**

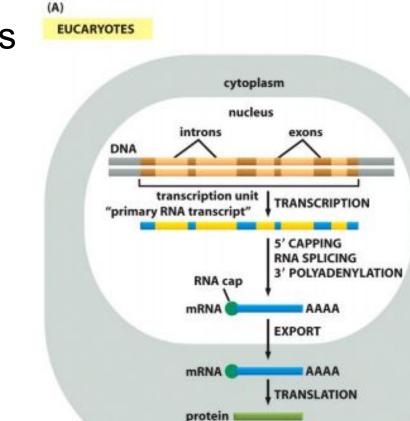
TAs: Sarah Cohen, Doo Young Lee, Erin Isaza, and Courtney Chen

# The Central Dogma 5 Biology **Fundamental!** 5′ H<sub>2</sub>N·

 $\bullet$ 

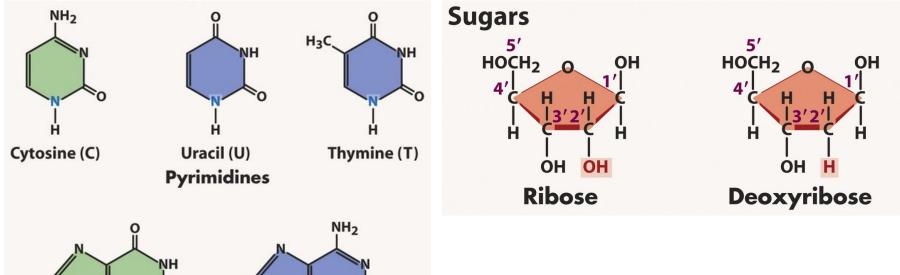


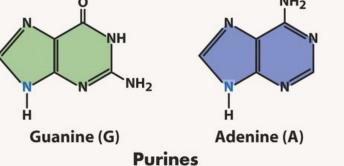


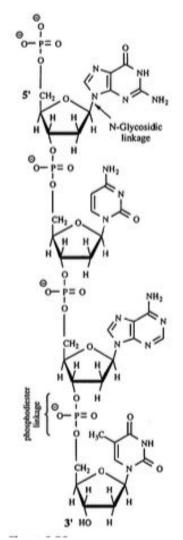


### **Prokaryotes and Eukaryotes**

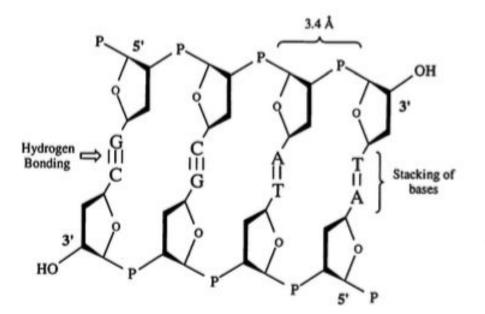
### Nucleic Acid Components



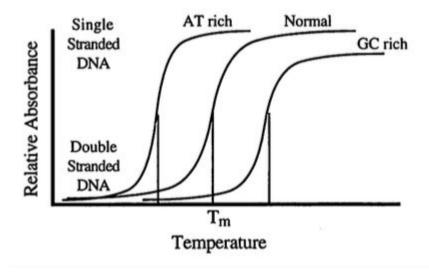


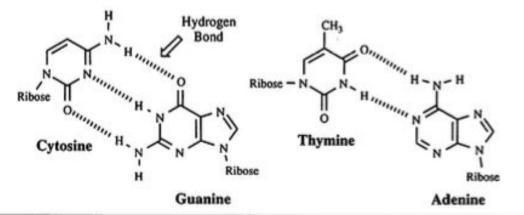


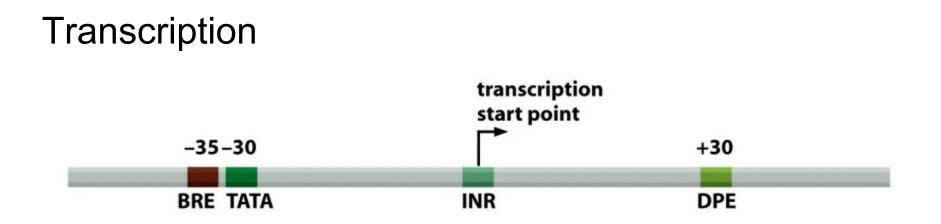




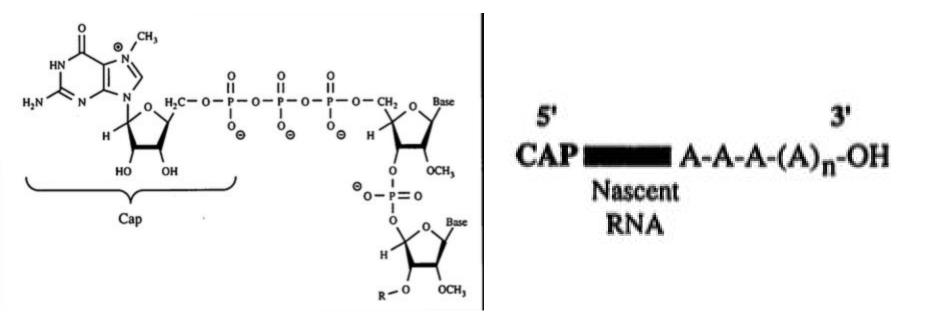
### **DNA Base Pair Interactions**

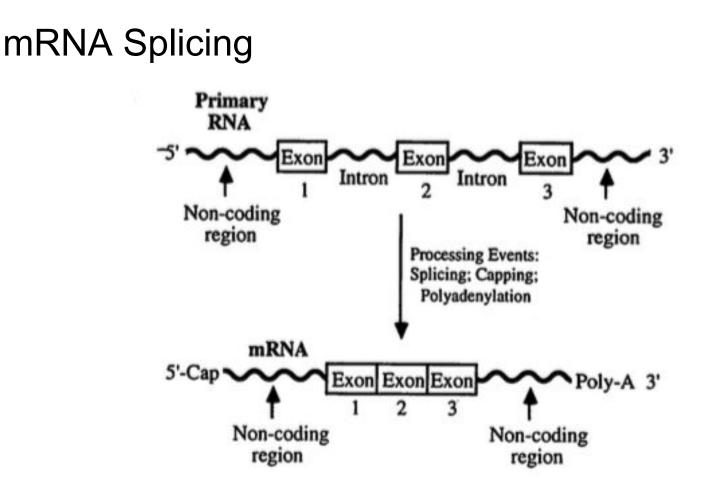






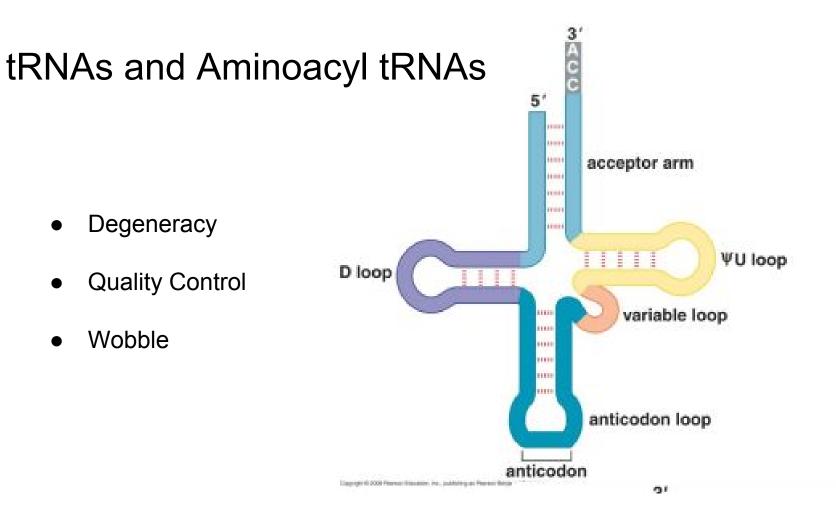
### **RNA** Modification



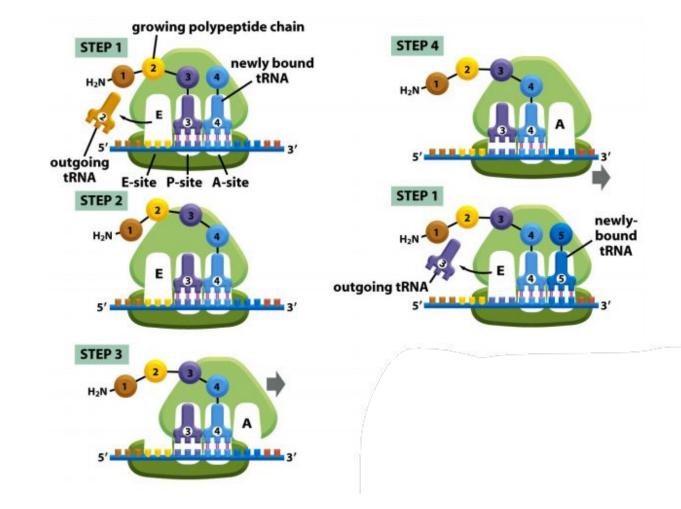


### Codons

		second	position		
_	U	с	A	G	
u	UUU Phe	UCU UCC Ser	UAU UAC	UGU UGC Cys	U C
	UUA UUG <sup>Leu</sup>	UCA	UAA* stop UAG* stop	UGA <sup>*</sup> stop UGG Trp	A G
c	CUU CUC	CCU CCC	CAU CAC His	CGU CGC	U C
	CUA CUG	CCA Pro CCG	CAA CAG Gin	CGA Arg	G U
	AUU AUC lle	ACU ACC	AAU AAC Asn	AGU AGC Ser	U . C
4	AUA AUG† Met	ACA Thr ACG	AAA AAG	AGA AGG Arg	A G
	GUU	GCU	GAU	GGU	U
G	GUC GUA Val	GCC Ala	GAC	GGC GGA Gly	C A
	GUG	GCG	GAG Glu	GGG	G



### Ribosomes



# **Proteins - The Basics**

- Built from amino acids
  - N to C terminus in synthesis direction
  - Amide bonds between carboxy group to amino group
  - Amino acid side chains determine types of interactions being made
- Structure and function are TIED TOGETHER
- Multiple types of ways that amino acids/domains can contribute
  - Structural support/stability
  - Positioning
  - Effector functions

# **Protein Structure Analysis - Fundamental Qs**

- What is the secondary/tertiary/quaternary structure looked at?
- Is the sequence examined part of an active interaction site or not?
  - If it is, what are the contacts being made/what residues are critical?
  - If not, is it structurally important?
- What is the conservation compared to a homologous protein (if there is one to be compared to) or mutant?
  - Where might they differ?
  - How important are the changes?
- How does the protein overall achieve its function?

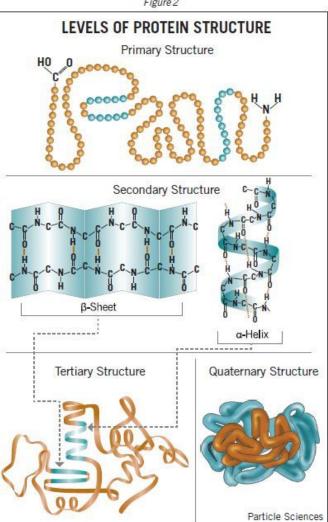
#### Figure 2



- Primary: more or less the amino acid sequence
  - Not 3D, not particularly informative on properties of prot.
- Secondary

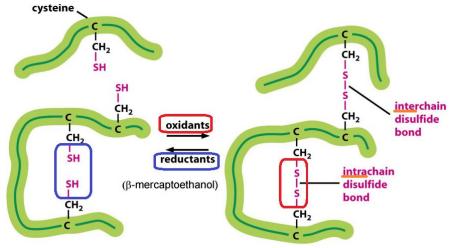
Ο

- Alpha helices Ο
  - 3.6 aa per turn
  - Side chains bristle
    - Spatial constraints
    - Common properties often same side
  - O to H hydrogen bonds ~4th aa below
    - Many bonds; highly stable
- Beta-pleated sheets Ο
  - Hydrogen bonds between strands not inside strands
  - C=O ---- H-N
  - Parallel vs antiparallel
    - Stronger when antiparallel b/c H bonds straight



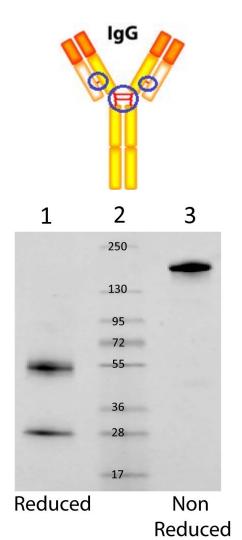
# Proteins - Structural Levels (cont.)

- Tertiary: OVERALL 3D shape
  - Conformation such that achieve lowest energy state/maximum stability
  - Hydrophobic side-chains buried inside (away from aqueous environment) and charged/polar will be exposed on surface (hydrophilic)
  - Disulfide bonds are formed by cysteines
    - True covalent bonds-- STRONG
    - Bridging = important stabilization of tertiary structure, links parts of chains
  - Salt bridges: ionic interactions via +/- aa side chains
- Quaternary: subunit interactions of protein to form final complex
  - Final shape via subunit H-bonds, salt bridges, disulfide bridges
  - Often formation of unique epitopes composed of multiple protein chains



# **Protein Stability**

- In a nutshell, proteins are held together by H-bonds mostly
  - Sensitive, multiple ways to mess w/3D structure
- Stressors:
  - pH, aqueous vs non-aqueous environment, salt/ions present, temp.
  - Essentially anything that would interrupt bonds
- Denature: unfolding of protein into misfolded/random shape
  - Result of loss of structure at secondary, tertiary, or quaternary
  - Can aggregate
  - Often will completely lose biological function compared to native form
- HELPFUL when dissecting structure!
  - Sizing w/ denaturing: reveals if individual or multimer
  - Sizing by disulfide reduction
  - Proteolysis to separate domains

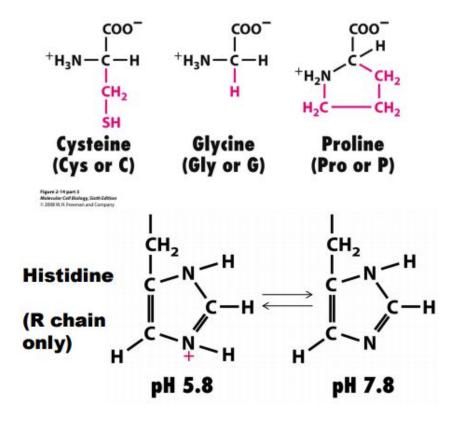


# **Proteins - Amino Acids**

- All amino acids contribute in some way but some are more characteristic
- Basic/acidic
  - tend to be chemically highly active
- Polar amino acids
  - In general oftentimes functionally important, not just structurally
    - Can be catalytic in nature
    - OR overall make important electrochemical interactions (i.e. more H-bonds)
- Nonpolar amino acids
  - Often associated w/hydrophobic regions, internalized, and add to stability
  - Ring structures can stack, which are strong stabilizing interactions

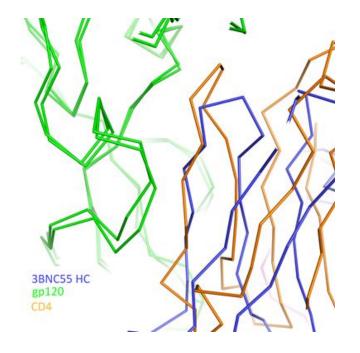
## Proteins - "Special" Amino Acids

- Cysteine R group has sulfhydrylgroup; forms disulfide bridges
  - Can DIMERIZE-- pairing w/another S
- Histidine pH dependent proton exchange (@pH 5.8, gains H at a nitrogen, loses @ pH 7.8); useful in enzyme catalysis
- Glycine smallest aa (H is its R group); can fit in tight parts of protein
  - Maximum flexibility
- Proline ring involves central carbon; rigidity from phi bonds results in "helix breaker", but in reality causes "kinks"/angles



# **Proteins - Conservation and Evolution**

- Examination of protein sequences in alignment can reveal a lot
  - If same, can potentially infer function of unknown protein to be similar to what it's aligned to
  - If different, can hypothesize which amino acids are causing changes to function
- HOWEVER, protein structures can be similar WITHOUT having similar sequences!
  - Structures can be mimicked numerous ways
- What's crucial:
  - Conservation of PARTICULAR amino acids for particular structural function can be more important <sup>CD4</sup><sub>3BNC55</sub> than OVERALL conservation
  - Prolonged conservation of certain residues over time despite drift can hint at necessity



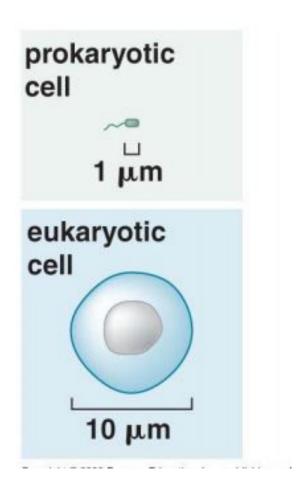
PSN	TKVDK	XVE	PKSCDK	- THTCPPCI	PAPELLG	SPSVFLFF	PKPKDT-
.:	*:*	*	*. ::	.: * .	: * * :	* . : :	: .:
CUTT	TOL MAN	EVENIONE		DUU	TIDOAL	owner	
2MT LL	DLKNK	EVENKKVI	<b>QDPKLQMGKKL</b>	PLAL-	- ILPQALI	-QYAGSGI	IL I LALEA

### Bacteria

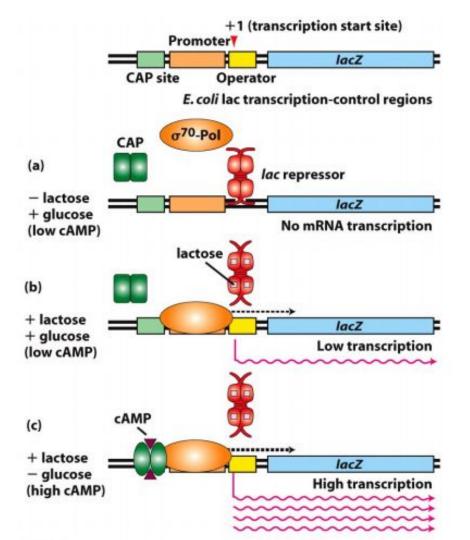
• Smaller, Faster, Simpler

• No separation of transcription/translation

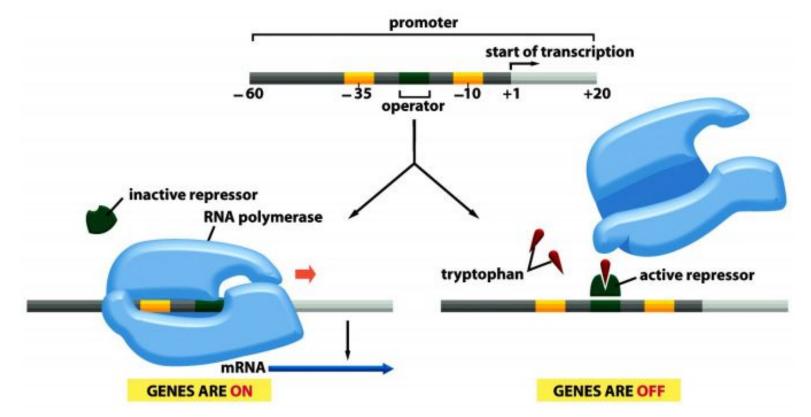
• Double-use of DNA



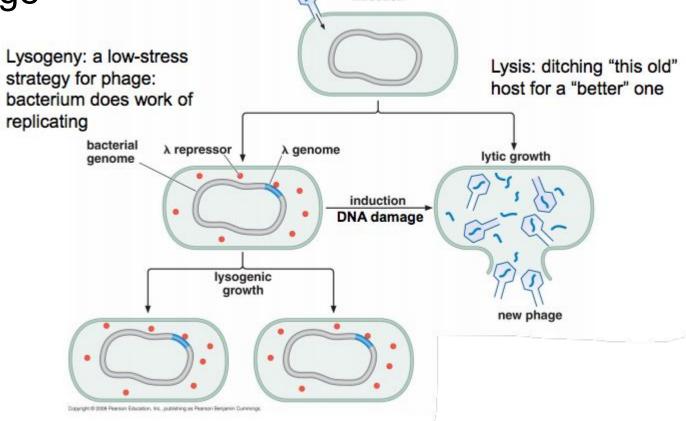
## Lac Operon



# Trp Operon



### Lambda Phage



infection